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Efficacy of Nano-Encapsulated Mandarin Oil using (Polyethylene Oxide/Polyacrylamide) and Gamma Irradiation against *Culex pipiens* Larvae

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#### ABSTRACT

Nano-encapsulation of essential plant oils may prove to be the most effective approach to overcome their application challenges against insect pests. Larvicidal properties of polyethylene oxide/polyacrylamide (PEO/PAAM) nanogels of mandarin essential oil (MEO) against Culexpipiens larvae were investigated. The nano-gel was prepared using an intramolecular crosslinking method initiated by gamma irradiation. The resulting mandarin essential oil (PEO/PAAm) nanogel (MEON) exhibited monodispersity, a relatively small size distribution, and a low negative surface charge of -1.1 mV, suggesting enhanced stability and reduced electrostatic interactions. The 2nd instar larvae were more sensitive to MEON than the 3<sup>rd</sup> and 4<sup>th</sup> instar larvae at the lower doses of 10 and 15 ppm. The  $LC_{50}$  and  $LC_{99}$  for the MEON were (11.5 and 18.4 ppm) for 2<sup>nd</sup> instar larvae; (14.5 and 23.6 ppm) for 3<sup>rd</sup> instar larvae; and (15.3 and 33.8 ppm) for the 4<sup>th</sup> instar larvae, respectively, at 48 hours after treatment. These results shed light on MEON's physical attributes, surface qualities, and biological assay and suggest that it might be a useful substance for *Culex pipiens* management.

#### **INTRODUCTION**

Culex mosquitoes are known to spread important diseases to humans, for instance, elephantiasis by the transmission of Lymphatic filariasis and West Nile, Rift Valley fever, and encephalitis by the transmission of arboviruses (Madeira *et al.*, 2024). *Culex pipiens* is the most abundant mosquito worldwide and is widely distributed in urban and suburban regions (Villena *et al.*, 2024). Excessive use of chemical insecticides has caused environmental pollution and vector resistance (Hillary *et al.*, 2024). Essential oils (EOs) were reported as effective toxic materials against mosquito larvae (Vivekanandhan *et al.*, 2023). The United States Authority for Food and Drugs has classified orange oils as safe. The oil contains many constituents such as limonene, monoterpenes,  $\alpha$ -pinene,  $\beta$ pinene, terpinolene, and octanal (de Jesus Oliveira *et al.*, 2024). *Citrus reticulata*, a mandarin EO that is a member of the Rutaceae family, has been the subject of numerous attempts to

Citation: *Egypt. Acad. J. Biolog. Sci.* (F. Toxicology& Pest control) *Vol.16(1) pp51-61 (2024)* **DOI: 10.21608/EAJBSF.2024.345753**  demonstrate the insecticidal action of orange essential oils (EOs) against various insect pest species (Sayed *et al.*, 2020; Marouf and Harras, 2022). (Brah *et al.*, 2023).

Although EOs have a toxic action against insect pests, they should be formulated to avoid the vaporization of volatile chemicals to preserve their actions (Osanloo et al., 2022). The submicron emulsion of nano-encapsulated essential oils (EOs) has been suggested as a potentially effective pesticide formulation; these formulations improve the solubility of poorly water-soluble oils (Ibrahim, 2020). Many nanoencapsulations of EOs were reported as pesticide formulations against mosquitoes (Esmaili *et al.*, 2021). Nano-gels, as a class of nanoscale hydrogels, have attracted significant attention in recent years due to their unique properties and promising applications in various fields, including biomedicine and drug delivery. One of the key challenges in nano-gel synthesis is achieving a controlled and uniform size distribution, which is crucial for their performance and effectiveness in specific applications. Various methods have been explored for nano-gel synthesis, including chemical crosslinking, self-assembly, and radiation-induced crosslinking (Matusiak et al., 2020a). The ideal intramolecular crosslinking method for synthesizing nanogels has been highlighted (Rosiak et al., 2005). Because gamma irradiation can both start and control crosslinking reactions in aqueous polymer systems, it has become a potential process for creating nano encapsulation. This technique offers advantages such as simplicity, scalability, and the absence of the need for initiators or additional chemicals. Specifically, the use of gamma irradiation in the intramolecular crosslinking process has demonstrated potential in the production of nano-gels with a restricted size distribution, a high crosslinking density, and a semi-permeable membrane (Alshangiti et al., 2019). For crosslinked hydrogel (Ghobashy et al., 2021b) to be used as a template (Ghobashy et al., 2021a) and blend polymer (Ghobashy et al., 2017) hydrogel points to utilize as green renewable assets to protect the environment from negative effects (Madani et al., 2022). Previous studies from our laboratory have indicated that irradiation treatment is a useful method for these hydrogel points. In order to boost the larvicidal efficacy of mandarin essential oil and expand its application against *Culex pipiens*, the current trial intends to create a unique formulation of the oil.

#### MATERIALS AND METHODS

#### **Insects and Chemicals:**

Mosquito larvae were reared in the insectary of the nuclear reach center, Inshas, Egypt. The larvae were reared under optimum humidity  $(75 \pm 5\%)$ , temperature  $(25 \pm 2\circ C)$ , and a photoperiod of 12:12 light/dark hours. Mandarin essential oil was purchased from the "Binsib Company for Agriculture Development" in Cairo, Egypt. PEO was obtained from Loba-chemieindoaustranal Co., Mumbai, India. Acrylamide monomer (AAm) was obtained from Sigma-Aldrich Co. Hydrochloric acid (35.5 wt.%) and sodium hydroxide flakes were supplied from the market.

#### Gamma-Irradiation-Induced Polymerization of (PEO/PAAm) Nanogel:

The optimum intramolecular crosslinking technique is used in the particular instance of the synthesis of polyethylene oxide/polyacrylamide (PEO/PAAm) nanogel, with gamma irradiation catalyzing radical polymerization processes. The experimental procedure involves preparing a 0.5 wt% solution of the AAm monomer and a 0.1 wt% solution of PEO by dissolving 0.05 g and 0.01 g of PEO in 10 ml of distilled water. The resulting mixture is then subjected to sonication for 5 minutes, which promotes the mixing and dispersion of the components within the solution. The pH of the solution is adjusted to 1 using a 35.5 wt% hydrochloric acid (HCl) solution. The pH adjustment is crucial for the subsequent polymerization reaction. Following the pH adjustment, the solution is exposed to gamma irradiation, which serves as the initiator of radical polymerization reactions. The irradiation dose employed in this synthesis is 5 kGy. The synthesis is conducted using a 60Co source established at NCRRT, EAEA, in Cairo, Egypt. The gamma irradiation triggers the crosslinking of the polymers, leading to the formation of the PEO/PAAm nanogel.

#### Encapsulation of Mandarin Oil Inside (PEO/PAAm) Nanogel:

The pH of the resulting suspension solution of PEO/PAAm) nanogel was adjusted to 6 by adding a concentrated solution of NaOH (added slowly while stirring at 150 rpm, allowing the nanogel particles to agitate). The resulting clear solution (A) was taken to the ultrasonic bath (70 w) for 15 minutes at room temperature. For the encapsulation of mandarin oil inside the PEO/PAAm) nano-gel, the emulsion solution of oil (B) was prepared by dissolving 5 ml of oil with 1 ml of tween 80 in 10 ml of distilled water, stirring at 350 rpm for 10 minutes, and keeping at 4 °C for 24 hours. Then, both solutions (A) and (B) were mixed gradually with stirring for 10 minutes, and the pH of the solution was adjusted to 4 using HCl solution (35.5 wt. %). The pH adjustment is crucial for maintaining the stability and properties of the encapsulated mandarin oil within the nanogel. The resulting mixture now contains the encapsulated mandarin (MEO) essential oil within the PEO/PAAm) nanogel (MEON).

#### Characterization of (PEO/PAAm)/MEO Nano-gel (MEON): Transmission Electron Microscope (TEM):

The structure of the MEO nanogel was examined by transmission electron microscopy (H-7650, Hitachi, Japan); the prepared formulation was already diluted with distilled water before the examination. One drop of the MEON formulation was put on a film-coated copper grid before staining with a (2% w/v) phosphotungstic acid solution. The grid was allowed to dry for 10 minutes at ambient temperature before visualization under a transmission electron microscope.

#### Dynamic Light Scattering (DLS) and Zeta Potential:

The particle size (PS) and polydispersity index (PDI) were investigated using Zetasizer Nano ZS (Ver.6.12, Malvern Instruments Ltd., Worcestershire, England) using the dynamic light scattering technique at room temperature. MEON was previously diluted with distilled water (100-fold) before any measurements. The structure of the MEON was examined by transmission electron microscopy (H-7650, Hitachi, Japan); the prepared formulation was already diluted with distilled water before the examination. One drop of the EO-loaded formulation was put on a film-coated copper grid before staining with a (2% w/v) phosphotungstic acid solution. The grid was allowed to dry for 10 minutes at ambient temperature before visualization under a transmission electron microscope. **Bioassay Tests:** 

A bioassay was performed using second, third, and fourth instar larvae of *Culex pipiens* kept at a temperature of  $25 \pm 2$  °C and a relative humidity of  $75 \pm 5\%$ . The larvae were treated with different concentrations of PEO, PAAM, and MEO nanogel according to the standard protocol. Four concentrations of nanogel (0, 10, 15, 20, 25, and 30 ppm) were used. For each treatment, five replicates of twenty-five larvae were used. Mortality was recorded after 24- and 48-hours post-treatment.

#### **Statistical Analysis:**

Lethal concentrations were determined at the 95% confidence level using a probit regression line. LC<sub>50</sub> and LC<sub>99</sub> were calculated (Finney, 1971). The percentages of larval mortality were calculated for each concentration of the tested samples. Correction for control mortality was conducted using Abbott's formula. A one-way ANOVA followed by Tukey's multiple comparison tests (P < 0.05) was carried out to illustrate the significance of the tested samples from the control groups. All statistical analysis was conducted using the statistical package for social science (SPSS) software version 14.

#### **RESULTS AND DISCUSSION**

#### Characterization of (PEO/PAAm)/MEO Nano-gel:

Figure 1, shows a transmission electron microscopy (TEM) image of MEON particles. The nano-gel particles appear monodisperse, meaning they have a uniform size distribution as seen in the TEM image. The diameters of the nano-gel particles are reported to be in the range of 130–160 nm based on the scale bar in the TEM micrograph.



Fig. 1: Transmission electron microscopy (TEM) of MEON.

The monodispersity and relatively small nano-gel particle diameters suggest the nano-gel may be suitable for various biomedical applications. The hydrodynamic diameter and distribution of encapsulated mandarin oil were determined by the dynamic light scattering (DLS) technique (Fig. 2.a). To deliberate the size distribution profile and polydispersity of the synthesized nanogel of mandarin oil, the particle size distribution of the MEON particles, as measured using a particle size analyzer, is reported to be 205 nm. This indicates that the nanogel has a relatively uniform size distribution centered on this value.



**Fig. 2:** particle size analyzer (a) particle size distribution = 205 nm and Zeta potential = -1.1 mV (b) of MEON.

The zeta potential of the nanogel indicates that all of the samples were negatively charged. The absolute value of the zeta potential reached a maximum when the amount of nanogel was  $4.2 \times 10-2\%$  w/v. The zeta potential value of encapsulated mandarin oil was 23 mV (Fig. 2.b), which indicates the strong repulsion between particles and the increasing stability of the encapsulated oil.

The gamma irradiation technique enables the synthesis of nano-gels in a one-step, simple, and scalable manner. Water-soluble polymers can be intramolecularly crosslinked by radiation to create nano-gels (Matusiak *et al.*, 2020). Based on the research of Rosiak *et al.* (2005), the optimal intermolecular crosslinking technique uses high-energy, low-LET (linear energy transfer) ionizing radiation to cause aqueous polymers to crosslink. This method results in nano-gels with a high degree of crosslinking and a narrow size distribution (Ashfaq et al., 2021). Radiation-induced crosslinking of polymers can be used to create hydrogels and their nanoscale counterparts, nano-gels (Dispenza *et al.*, 2017). To develop botanical biopesticides for use in plant protection, numerous studies have attempted to create EO-based nano-emulsions (Campolo *et al.*, 2020). Nano-emulsions are kinetically stable systems with droplets ranging in size from 50 to 200 nm (Lakshmayya *et al.*, 2023). Nano-encapsulation using polymers is a well-known method for the preservation of essential oils. It offers plenty of benefits, including improved water solvency, compelling assurance against degradation, the avoidance of volatile component evaporation, and tightly controlled and targeted release (Lammari *et al.*, 2020).

#### **Bioassay**

Figure 3, shows the mortality of 2nd instar larvae of C. pipines at 24 h after treatment with MEON concentrations. Significant mortalities were obtained in the cases of 15, 20, and 25 ppm of MEON compared to concentrations of 0.0 and 10 ppm ( $F_{(4, 24)} = 2464.2$ , P<0.05). **Fig. 3:** Mortality percentages of different concentrations (ppm) of MEON against 2<sup>nd</sup> instar



larvae of C. pipines at 24 and 48 h post treatments.

While after 48 hours, mortality increased gradually by increasing the concentration levels ( $F_{(4,24)} = 29.5$ , P<0.05). The higher mortalities (96.0, 99.2, and 99.9%) were observed at the tested concentrations (15, 20, and 25 ppm), respectively.

Similarly, higher mortality (99.7%) was obtained for 3rd larvae treated for 25 ppm concentration after 24 h treatments than the mortalities recorded for the other tested concentrations ( $F_{(4,24)} = 461.6$ , P< 0.05) (Fig. 4). While within 48 h after treatment, the mortality at 10 ppm concentration was still drastically low compared to 15, 20, and 25 ppm ( $F_{(4,24)} = 682.5$ , P< 0.05). Obtained data revealed that the mortalities of the 3<sup>rd</sup> larvae were lower than those of the 2<sup>nd</sup> larvae.



**Fig. 4.** Mortality percentages of different concentrations (ppm) of MEON against  $3^{rd}$  instar larvae of *C. pipines* at 24 and 48 h post treatments.

As shown in Figure 5, the mortality percentages of treated 4th larvae at 20 ppm were relatively lower (50.4%) compared to 25 ppm (99.56%) at 24 h after treatment (F (4.24) = 433.1, P< 0.0005), while they were highly increased to 68.0% at 20 ppm, and the same percentage of mortality (99.56%) was obtained at 25 ppm for 48 h after treatment ( $F_{(4,24)} =$ 249.3, P<0.0005). Moreover, the percentage of mortality increased from 7.2 to 15.2 for 24 and 48 hours at 10 ppm, respectively. Indeed, many trials indicated that orange EOs have toxic effects on mosquito vector diseases, for instance, anopheles mosquitoes (Umar et al., 2024), Culex quinquefasciatus (Suwansirisilp et al., 2013), Culex pipiens (Michaelakis et al., 2009), and Aedes albopictus (Li et al., 2023). The obtained results indicated that the MEON concentration (25 ppm) exhibited more than 99 % mortality in C. pipiens instars larvae, which was drastically lower than those recorded for bulk orange EOs in the previous studies, which ranged from 300 to 1000 ppm (M. Azmy et al., 2019a) (Manimaran et al., 2012). It was demonstrated that mandarin EO inhabited the acetylcholine esterase, carboxyl esterase, acid phosphatase, and alkaline phosphatase of C. pipiens larvae (Badawy et al., 2018). It was found that over 2000 secondary metabolites have been identified from orange essential oils, with the most important active ingredients being diterpenes, monoterpenes, terpenes, and sesquiterpenes, which may refer to the toxicological effects of the orange EOs against insect pests (Brah et al., 2023). The current study is comparable with several studies on nanoemulsions based on EOs as effective insecticides (Sogan et al., 2023; Abdel-Baki et al., 2021). Because the EOs have destitute water solubility, larvicides utilized to manage mosquito larvae ought to be dissolvable in water since they are aquatic life forms. To bypass this natural barrier, EOs should be formed into nanoemulsions (Sharifiyan *et al.*, 2024). The obtained results in the concentration mortality bioassays were satisfactorily described by the probit model (Fig. 6).







**Fig. 6:** Mortality-responses of *C. pipines* larvae  $(2^{nd}, 3^{rd} \text{ and } 4^{th})$  for different concentrations (ppm) of MEON at 48 h post treatments.

The LC<sub>50</sub> was (11.5, 14.8 and 15.3 ppm) for the 2<sup>nd</sup> instar, 3<sup>rd</sup> and 4<sup>th</sup> instar, respectively (Fig. 6). Also, the LC<sub>99</sub> was (18.4, 23.6 and 33.8 ppm) for the 2<sup>nd</sup> instar, 3<sup>rd</sup> and  $4^{\text{th}}$  instar, respectively. The results indicated that LC<sub>50</sub> the fourth instar was more resistant to treatment, but the second instar was much more vulnerable to MEON (no overlap in 95% fiducial limits), according to the  $LC_{50}$  and  $LC_{99}$  values. As shown in Figure 6, a concentration-dependent association between MEON and larval mortality was demonstrated via regression analysis. Insecticidal properties of orange oils were reported for larvicidal activity against C. pipiens in a previous study that suggested it might be because of transanethole, the main component of the EO (Kamaraj et al., 2023). The orange EO nanoemulsion's larvicidal action could be attributed to limonene, the main ingredient, which Saad (2013) showed has insecticidal qualities. Our results are in line with several studies on nanoformulations that use essential oils (EOs) as effective pesticides (Duarte et al., 2015; Barradas and de Holanda e Silva, 2021). Our results are consistent with those of Sanei-Dehkordi et al. (2022), who discovered that the LC<sub>50</sub> and LC<sub>90</sub> of nano orange EO for the third larvae of Anopheles stephensi and Culex quinquefasciatus, respectively, were the low concentrations (6.63 and 12.29 µg/mL) and (4.9 and 16.4 µg/mL), respectively. The results we obtained are in agreement with those reported by Theochari et al. (2020), who observed that a concentration of 27.4 ppm of nanoemulsion orange EO was reported as an  $LC_{50}$  against C. pipines larvae in their third instar. Based on our research, it appears that the MEON could be the most effective way to address the issues with using EO against *C. pipiens*. It does this by preventing rapid evaporation and degradation in the environment, offering a well-soluble formulation, and lowering the dosage required. To guarantee this novel material's safety for the environment and non-target creatures, more investigation is necessary for the risk evaluation of the material. Studying field applications and challenges related to economic feasibility is also necessary.

#### CONCLUSIONS

Orange essential oils have become one of the major bio-pesticides against mosquito vector diseases due to their known toxicity and their environmental benefits. However, it is essential to preserve the orange essential oils by maintaining their stability against environmental conditions since they are chemically unstable and can degrade when exposed to air, light, moisture, and mild temperatures. High doses and poor water solubility are also significant considerations when using essential oils. The designated emphasis in this trial is using nano-gel as a green encapsulation of mandarin essential oil in the hope of revolutionizing their application challenges. The aqueous polymer polyethylene oxide/polyacrylamide was crosslinked by gamma irradiation, producing polyethylene oxide/polyacrylamide nano-gels as a result. Mandarin essential oil (MEO) was used to dissolve the solution and create mandarin essential oil nano-gel (MEON), which has a restricted size distribution and a high degree of cross-linking. The resulting MEON presented spherical shapes with homogeneous surfaces. The average size of MEON particles ranged from 130 to 160 nm; a low negative zeta potential of -23.0 mV indicated high stability of the emulsion. The results demonstrated that the dose of 25 ppm of MENO exhibited 99.9% mortality in the three instar larvae of C. pipiens. Further research on the risk assessment of MENO is required for its safety properties.

Ethics Approval and Consent to Participate: Not applicable.

Competing interests: The authors declare no conflict of interest.

**Authors Contributions:** I hereby verify that all authors mentioned on the title page have made substantial contributions to the conception and design of the study, have thoroughly reviewed the manuscript, confirm the accuracy and authenticity of the data and its interpretation, and consent to its submission.

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